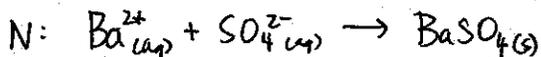
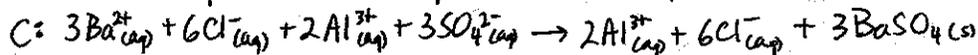
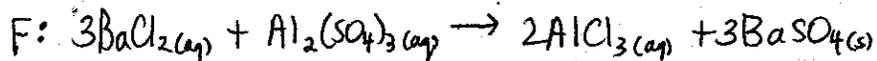


F, C, N EQUATIONS WORKSHEET

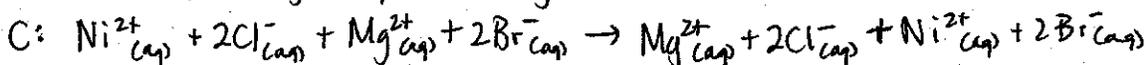
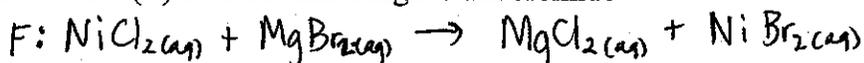
Name: KEY

For each of the following write a *formula equation*, a *complete ionic equation*, and a *net ionic equation*. Include states (*s* or *aq*) and balance each equation.

1. Barium chloride and Aluminum sulfate

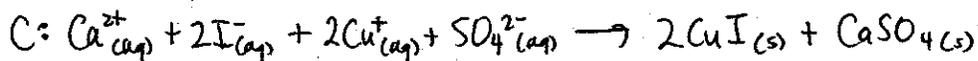
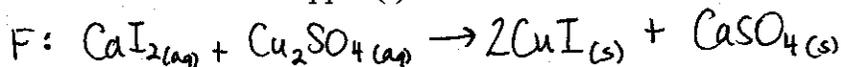


2. Nickel (II) chloride and Magnesium bromide



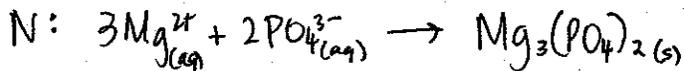
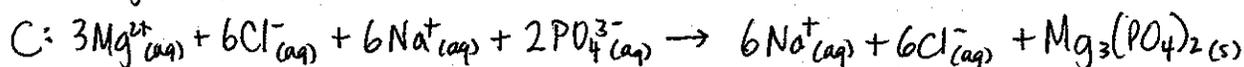
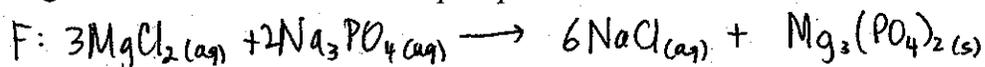
N: none

3. Calcium iodide and Copper (I) sulfate

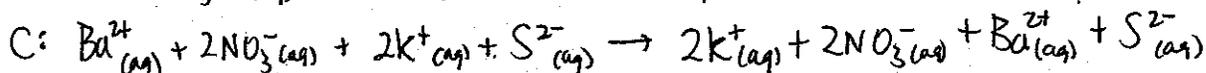
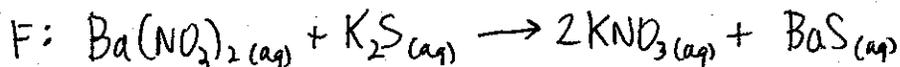


N: same as complete ionic

4. Magnesium chloride and Sodium phosphate



5. Barium nitrate and Potassium sulfide



N: none